



醫用磁共振學MRM Chemical Exchange Saturation Transfer - CEST

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本週課程內容 <http://www.ym.edu.tw/~cflu>

- 化學交換飽和轉移 Chemical exchange saturation transfer (CEST)

- CEST: from basic principles to applications, challenges and opportunities - E. Vinogradov, A. Sherry, R. Lenkinski, *J Magn Reson.*, (2013) 229: 155.
- Chemical Exchange Saturation Transfer: What is in a Name and What Isn't? - Peter C. M. van Zijl and Nirbhay N. Yadav, *Magnetic Resonance in Medicine*, (2011) 65:927-948.

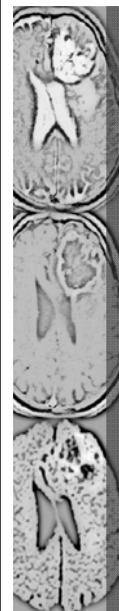
<http://www.ym.edu.tw/~cflu>, Textbook: MRI The Basics, Hashemi et al.

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化學交換飽和轉移原理

Principles of CEST



Why CEST-MRI?

- Conventional MRI gives T1, T2, and proton density weighted images.
- The most common contrast agent, Gd-chelate, shortens the relaxation time of the free water protons.
 - These agents are not selective, and distribute uniformly throughout the extracellular space after intravenous injection.
 - These agents are often toxic.

CEST provides a molecule-specific contrast for MRI.

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Principles of CEST

- A slowly exchanging MR-active nucleus, typically a proton, possessing a chemical shift distinct from water is selectively saturated and the saturated spin is transferred to the bulk water via chemical exchange.

→ New contrast with the normal T1-relaxation time

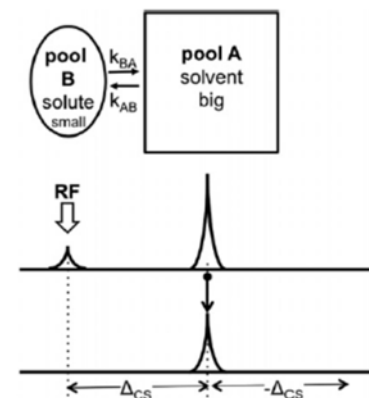
→ Indirect measurement of (metabolic) exchange-processes

Introduced by Balaban and co-workers as a new class of MRI contrast in 2000.

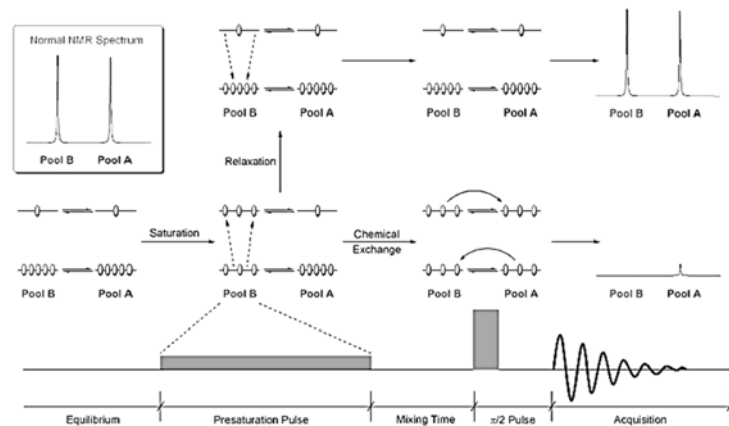
Schematic of CEST

- Pool A (solvent) is in exchange with pool B (solute).
- Pools A and B have distinct chemical shifts, with the difference of Δ_{CS} .
- RF is applied on-resonance with pool B resulting in saturation transfer and signal decrease of pool A.

Solute:
μM to mM range.



Scheme of CEST



M Woods, Chem Soc Rev 2006, 35(6): 500-511.

Essentials for CEST

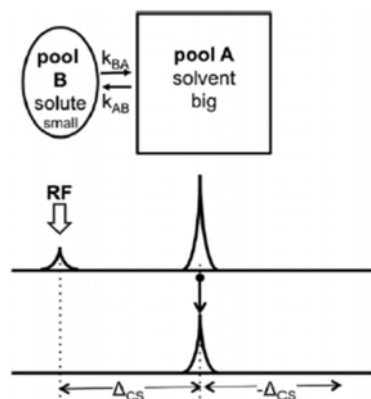
- A pool with target protons that will exchange to water
- Frequency-specific RF Pulse to saturate the protons
- Mathematical processing of the data

Exchangeable Protons

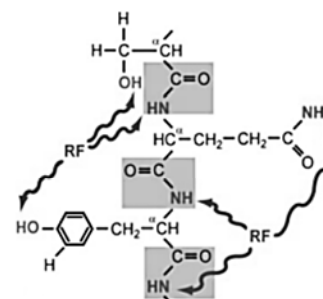
- Must possess exchangeable protons or molecules.
- The exchange rate k_{ex} must be sufficiently fast
- the chemical shift Δ_{CS} has to be big enough $\Delta_{CS} > k_{ex}$

The exchange rate:

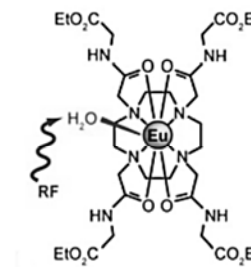
$$k_{ex} = k_{BA} + k_{AB}$$



Types of CEST Agents



ATOM (PROTON) EXCHANGE



MOLECULAR EXCHANGE



COMPARTMENTAL EXCHANGE

Types of CEST Agents

- Proton exchange
 - Endogenous and exogenous diamagnetic CEST (diaCEST)
 - Some paramagnetic CEST (paraCEST)
- Molecular exchange
 - paraCEST
 - Hyper-polarized gases (**hyperCEST**)
- Compartmental exchange
 - Liposomes (lipoCEST) and hyperCEST

diaCEST

- The chemical types of proton exchange groups are largely confined to
 - -NH
 - -NH₂
 - -OH
- The chemical shift of the diaCEST agents is typically within 5 ppm (relatively small shift) from water.
 - In the environment of 9.4T, only the exchange rates of $2 \times 10^3 \text{ s}^{-1}$ or slower are needed for CEST.
- Increase the number of the exchanging groups per agent can increase the CEST effect → polymers or dendrimers.



diaCEST

- Involves both endogenous and exogenous proton exchange types.
- Exogenous injection of the agent
 - Glucose imaging (glucoCEST)
 - pH imaging using iopamidol



paraCEST

- Exogenous paramagnetic lanthanide (III) complexes that exhibit large hyperfine shifts (on the order of 50–700 ppm).
- Pre-saturated targets
 - The highly shifted bound water protons
 - The ligand's amide
 - Hydroxyl protons
- Usages in reporting biological indices
 - pH, temperature, lactate, glucose concentration, enzyme activity, cell-labeling.



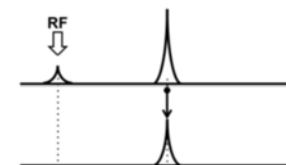
paraCEST

- Larger chemical shift allows a wide range of exchange rates.
- The fast exchange rates should allow detection of much lower concentrations of paraCEST agents in comparison with diaCEST.
- The selective RF pulse is applied far from the free water resonance frequency, which reduces direct RF saturation.



The Pre-saturation Pulse

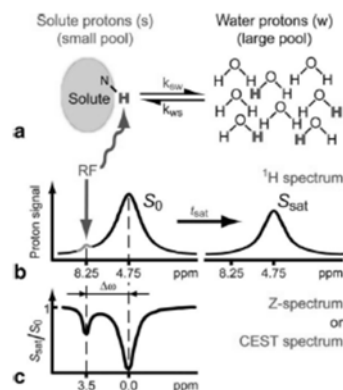
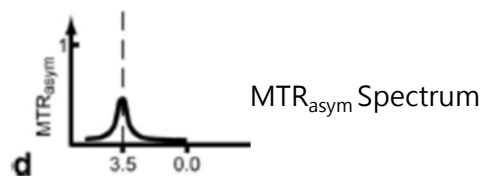
- Saturate exchangeable protons with RF-pulse
- → the protons exchange with water-protons and the water signal gets lower.
- → if t_{sat} longer than the exchange, the water-signal gets lowered visibly because the saturation builds up



Z-Spectrum

- Normalized water Spectrum (S_{sat}/S_0)
- Magnetization transfer asymmetry

$$MTR_{asym} = \{S_{sat}(-\Delta\omega) - S_{sat}(\Delta\omega)\} / S_0$$



Two-Pool Model

- Assume CEST contrast = proton transfer ratio (PTR)
- Analytical solution:

$$PTR = x_s \cdot \alpha \cdot k_{sw} \cdot T_{1w} \left(1 - e^{-t_{sat}/T_{1w}}\right),$$

$$x_s = \frac{[\text{exchangeable proton}]}{[\text{water proton}]} = \frac{k_{ws}}{k_{sw}}$$

k_{sw} = exchangerate saturation-pool \rightarrow water

α = saturation efficiency

T_{1w} = T_1 of water

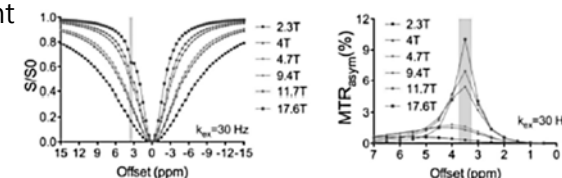
t_{sat} = saturation time

Factors of CEST

- The magnetic fields
- The Pulse-Sequences
- Imaging issue
- Different Z-Spectra of molecules

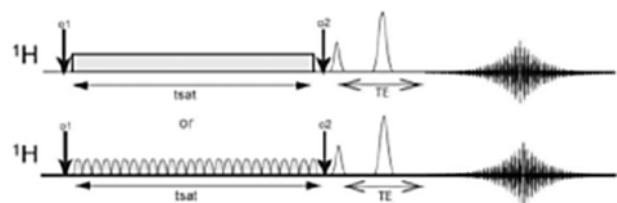
The Magnetic Fields

- From experiments: $B_0 \uparrow \rightarrow T_1 \uparrow$ & $T_2 \downarrow$
- Δ_{CS} is proportional to strength of the main-field
- higher Δ_{CS} allows agents with faster exchange rate
- long T_1 means slower recovery from saturation \rightarrow larger CEST-contrast
- Increase B_0 increases the sensitivity \rightarrow CEST is suitable for the high-field measurement



The Pulse-Sequences

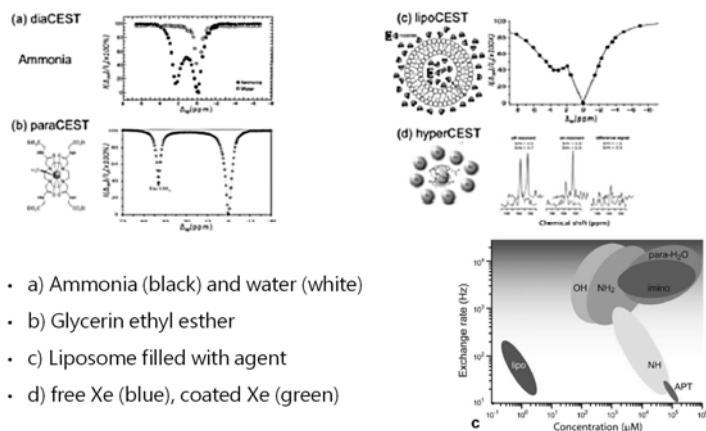
- One saturation-pulse before the T1-image sequence
 - one long saturation pulse
 - repetition of many short pulses
- To minimize the recovery of from the saturation \rightarrow shorter pulse and dephasing after every taken slice



Imaging Issues

- Problems while taking CEST-MRI:
 - Time: $t_{\text{sat}} > 2 \text{ sec}$ \rightarrow time for measurement $> 9 \text{ min}$ per saturation (need more for imaging).
- For faster measurement
 - limited offset sampling
 - reduce sampling points in k-space
- Desaturation: Spins flip back from saturation and signal gets lower.
- Solutions:
 - shorter pulse and dephasing after every taken slice
 - reduce taken time for measurement

Different Z-Spectra of molecules



- a) Ammonia (black) and water (white)
- b) Glycerin ethyl ester
- c) Liposome filled with agent
- d) free Xe (blue), coated Xe (green)

Summary

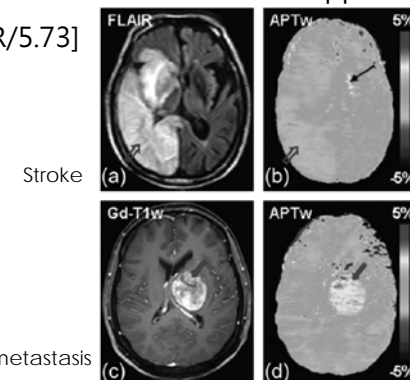
- CEST-MRI measures indirect exchange of protons/molecules \rightarrow (indirect) measurement of processes in the tissue
- Suitable for high-field measurements, 7T or 9.4T
- Further development with more exchange-mechanism and further contrasts arise

Applications of CEST

- The CEST effect can be used to image important physiological parameters, such as pH and metabolite levels.
- The technique can be applied for variety of ailments and metabolic disorders
 - cancer,
 - ischemia,
 - cartilage degeneration

Amide Proton Transfer (APT)

- The maximum APT effect is observed at 3.5 ppm from water
- $\text{pH} = 9.4 + \log[\text{APTR}/5.73]$



THE END

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